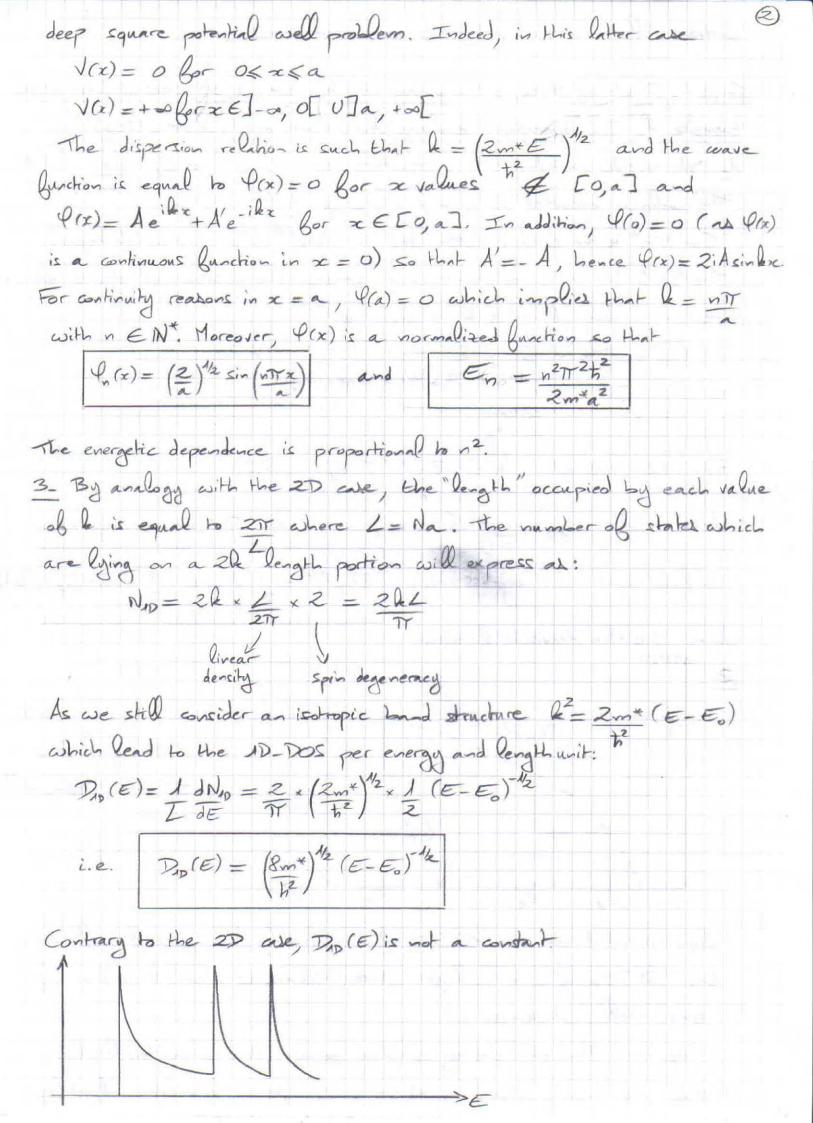
Note here that the energy spacing between the quantized levels increases when moving toward excited states as expected from the infinitely



Exercise 2 - Ionization energical 1. The situation is identical to that of the hydrogen atom except the fact that acceptors (or donors) lie in a layer characterized by a large dielectric constant. The ionization energy will express as  $E_i = \frac{t^2}{2ma_o^2} \frac{m^* 1}{E_i^2 n^2}$  (2.1) cohich resumes to  $E_i = 13.6 \, \text{m} \pm \frac{1}{E_i^2} \, \text{n}^2$  is we want to get energies in eV.  $m_i^* = m$  is the reduced mass of the carrier under consideration and  $E_r = E/E_0$ is the relative dielectric constant of the semiconductor and n is the principal quantum number of the orbital of interest (as is the hydrogen Bohr radius, a = to = 0.53 Å). The ionization energy is the energy difference between the n = 1 energy state (n=1, energetic level of the acceptor) and the top of the valence band (n=+00). We will thus have Einstein = 13600 x 0.53 x 1 ~ 50.3 meV The various peaks seen in figure 1 correspond to transitions between the level n=1 and levels n=2 to 4.  $\Delta E_{12} = E_{ioritation} \times (1 - \frac{1}{Q}) = \frac{3}{4} E_{ioritation} = 38.2 \text{ meV}$ DEIS = Eignitation × (1-1/3)= 8 Eignitution = 45.2 meV DEAG = Eionization × (1-1) = 15 Eionization = 47.7 meV Graphically, we deduce the bollowing energies: 33.8 meV, 39 meV, 42.9 meV and 51.9 meV We see here the limits of the hydrogen-like model as the effective masses change considerably in the vicinity of impurities n= 1 (acceptor tesel)

DER () = absorption lines > b. is partially ionized N=3 N=40 N=40

Such a measurement can only be carried out at low temperature, i.e. when donors and acceptors are neutral. It will also limit the impact of phonons. The decrease of the absorption coefficient for energies superior to the broad peak at a 51 meV results from the strong decrease of the transition probability away from the top of the valence band. The limited extent of the acceptor in momentum space is then responsible for the declining absorption at photon energies greater than the Linding energy of the acceptor state.

Exercise 3 - Activation energies

Data are plotted on a semilogarithmic scale.

Ly @ linear slopes => exponential dependence

@p(12cm) 15. 1/4

3 when Tirerealer, p decreases

D+D+3 => exp(E)

Low temperature range (curve A)

P1 = 0.05 12 cm 1/4 = 3×10-3 K-1

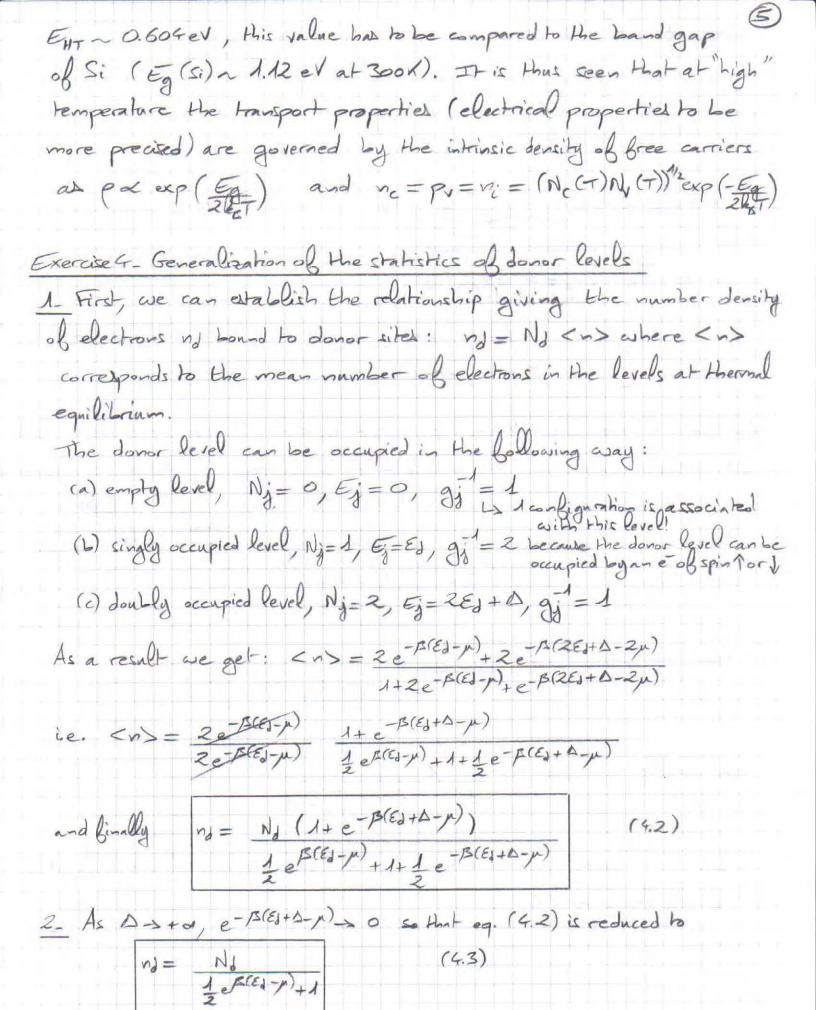
Cz = 852 cm 1/Tz = 1.2×10-2 K-1

ELT = & (= - =) ln(P1) ~ 7.78 x 10-21 J i.e. ~ 48.6 me.

This value has to be compared to the ionization energy of phosphorus atoms in silicon (donor state) ~ 45 met given in the lecture (idem in the article by G.L. Pearson and J. Bardeen, Phys. Rev. 75, 865 (1949)).

High temperature range (curve A)

P3 = 2.1 × 10-3 S2 cm 1/ = 5 × 10-4 K-1 P4 = 0.07 S2 cm 1/ = 3 × 10-3 K-1



3- As  $\Delta \rightarrow 0$ , (4.2) can be simplified in the following any:  $n_{d} = \frac{1 + e^{-\beta(\xi_{d} - \mu)}}{\frac{1}{2}e^{\beta(\xi_{d} - \mu)} + 1 + \frac{1}{2}e^{-\beta(\xi_{d} - \mu)}}$ Nd

finally lends to (4.4).